

cis isomer corresponding to 1c; the reaction time, however, was longer (6h).

TABLE(4,5)

Compd	Time (h)	Yield (%)	M.p. (°C)	IR (nujol) (cm ⁻¹)	NMR (CD ₃ COCD ₃) (δ)
<u>2a</u>	0.5	66	114	3310(NH), 1740 and 1720(CO)	1.25, 1.28(6H, two t, CH ₃), 4.22(4H, q, CH ₂), 6.9-7.4(4H, m, ar), 7.86 (1H, s, CH=), 8.1(1H, broad s, NH)
<u>2b</u>	0.25	58	168	3300(NH), 2220 (CN), 1740(CO)	1.30(3H, t, CH ₃), 4.29(2H, q, CH ₂), 6.8-7.5(4H, m, ar), 7.66(1H, s, CH=), 8.3(1H, broad s, NH)
<u>2c</u>	2	25 ⁽⁶⁾	131	3310(NH), 1730 (CO)	0.91(3H, t, CH ₃), 3.96(2H, q, CH ₂), 7.0-7.45(10H, m, ar and CH=), 8.0 (1H, broad s, NH)

Since the treatment of 1-chlorohydrazone with triethylamine is a well known procedure for generating in situ nitrile imines, it seems reasonable to posit that intermediates 2 are initially formed in the course of the above reaction. These intermediates could then evolve according to one of the following alternative pathways; a) intramolecular cycloaddition to the olefinic function to give strained tricyclic species, which should rearrange quickly to the final products; b) 1,7-electrocyclic ring closure, followed by prototropic shift.

Work is in progress to provide further mechanistic informations, particularly by varying the olefinic substituents.

References and notes

- (1) T. Tsuchiya, J. Kurita, H. Igeta, Chem.Comm., 640 (1974).
- (2) J. Kurita, T. Tsuchiya, Chem.Comm., 936 (1974).
- (3) Compounds 1a,b show for the olefinic protons $J = 16$ Hz. In the case of 1c, the signal of these protons is masked by that of the aromatics; however, the assigned structure was proved to be correct through the independent synthesis of the cis isomer ($J = 12$ Hz).
- (4) Satisfactory elemental analyses were obtained for all the compounds listed.
- (5) NMR spectra were taken at 60 MHz with tetramethylsilane as the internal standard. Melting points are uncorrected.
- (6) After chromatography on silica gel column.